B091 Some chemical constituents of Scilla natalensis and Urginea altissima (Hyacinthaceae)

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The Hyacinthaceae family is one of the two most widely used plant families by the Zulu of KwaZulu-Natal (1). Recent investigations in our laboratory of several members of the three Southern African sub-families of the Hyacinthaceae have yielded a range of novel homoisoflavanones, nor-triterpenoids, bufadienolide glycosides, cholestane glycosides, chalcones, and benzopyranones. These include a novel bufadienolide glycoside, urginin,

 $(3\beta$ -O-{α-L-rhamnopyranosyl-[(1 \rightarrow 4)-β-D-glucopyranosyl] 1 \rightarrow 3)-α-L-rhamnopyranoside}) which precipitated out from the methanol extract of the bulbs of Urginea altissima, and the novel trisnortriterpenoid, (23S)- 17α , 23-epoxy- 3β , 24 ξ -dihydroxy-27, 28, 29trisnor-lanost-8-ene, 1, and bisnortriterpenoid, (22R, 23S-17 α ,23-epoxy-3 β ,22,24 ξ -trihydroxy-27,28-bisnor-lanost-8-ene, 2, isolated from the dichloromethane extract of the bulbs of Scilla natalensis. Compounds 1 and 2 were isolated using column chromatography over silica gel and structures were determined using 2D-NMR techniques and LC-MS/MS methods for urginin and GC-MS for compounds 1 and 2.

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References: 1. Pohl,T.S. et al. (2000) Curr. Org. Chem. 4: 1287-1324.

B092 Chemical constituents of the Zulu medicinal plant Galtonia princeps (Hyacinthaceae)

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Galtonia princeps is a member of the Ornithogaloideae subfamily of the Hyacinthaceae family. Bulbs of this plant are used for magical purposes by the Zulu people of KwaZulu-Natal. An investigation of the chemistry of the bulbs of this species has yielded a novel cholestane glucoside from the methanol extract, and a homoisoflavanone from the dichloromethane extract, as shown below. This is the first report of the isolation of a homoisoflavanone from outside the Hyacinthoideae sub-family of the Hyacinthaceae (1). Compounds were purified by means of column chromatography over silica gel and structures were determined using 2D-NMR and MS techniques. The identity of the sugar was determined using an acid hydrolysis and identification of the sugar obtained.

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References: 1. Pohl, T.S. et al. (2000) Curr. Org. Chem. 4: 1287-1324.